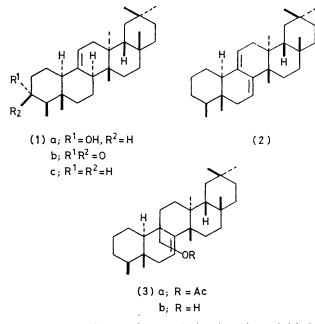
Terpenoids and Related Compounds. Part 25.¹ Structures of the 25-Nortriterpenoids Putrol and Putrone

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The suggested structures of putrol (1a) and putrone (1b) have been supported by the conversion of putrone into D: A-friedo-oleana-7,9(11)-diene (2). The latter has also been synthesised from 25-acetoxy-D: A-friedo-olean-7-ene (3a).

SESHADRI and his co-workers ² isolated an alcohol (putrol) and its corresponding ketone (putrone) from the leaves of *Putranjiva roxburghii*. From physical data, especially mass spectrometric fragmentation patterns, putrol and putrone were suggested ² to be the 25-nortriterpenoids (1a and b), respectively. However these structures were



not correlated with any known derivative of D: A-friedooleanane. In this paper we confirm these structures by correlating them with that of a known D: A-friedo-oleanane derivative.

Putrone (1b), isolated from the leaves of *P. roxburghii*, was subjected to Huang Minlon reduction to yield 25nor-D: A-friedo-olean-9(11)-ene (1c), which on oxidation with selenium dioxide furnished 25-nor-D: A-friedooleana-7,9-(11)-diene (2).

The diene (2) was also prepared from 25-acetoxy-D: A-friedo-olean-7-ene (3a), the synthesis of which we have previously described.³ The acetate (3a), on treatment with lithium aluminium hydride, gave the corresponding unsaturated alcohol (3b), which on refluxing with lead tetra-acetate and copper(II) diacetate in benzene under nitrogen furnished the diene (2), identical with the sample obtained from putrone (1b).

This not only confirms the structures of putrol (la) and

putrone (1b) but settles their stereochemistry at all asymmetric centres except C-3 in putrol (1a).

EXPERIMENTAL

N.m.r. spectra were recorded in deuteriochloroform at 100 MHz using tetramethylsilane as internal standard. Optical rotations were measured in chloroform. Light petroleum used had b.p. $60-80^{\circ}$ unless otherwise stated.

25-Nor-D: A-friedo-olean-9(11)-ene (1c).—A mixture of putrone (1b) (0.4 g) (isolated from the leaves of *P. roxburghii* following the method of Seshadri and his co-workers ²) and 99% hydrazine hydrate (5 ml) in diethylene glycol (40 ml) was refluxed for 1 h. Potassium hydroxide (0.4 g) was added and the mixture was refluxed for a further 1 h. The condenser was then removed and the mixture heated until the inner temperature reached 190 °C. The condenser was replaced and the mixture was refluxed for a further 12 h. Usual work-up furnished a solid (0.4 g) which was chromatographed over silica gel (20 g). Elution with light petroleum yielded a solid (0.36 g) which was crystallised from chloroform-methanol to give the pure *alkene* (1c), m.p. 170—174°, $[\alpha]_{\rm D}$ 2.2° (Found: C, 87.2; H, 12.0. C₂₉H₄₈ requires C, 87.4; H, 12.2%), v_{max}. 850 cm⁻¹ (trisubstituted double bond), δ 0.62—1.52 (4-, 5-, 13-, 14-, and 17-Me and 20-Me₂) and 5.23br (1 H, H-11), *m/e* 396 (*M*⁺), 381, 257, 243, 229, 205, 191, and 123.

Oxidation of (1c) with Selenium Dioxide. 25-Nor-D: A-friedo-oleana-7,9(11)-diene (2).—The hydrocarbon (1c) (0.2 g) and selenium dioxide (0.2 g) in glacial acetic acid (60 ml) were heated on a steam-bath for 1 h. The mixture was filtered and the filtrate on usual work-up furnished a semisolid mass (0.2 g) which was chromatographed over silica gel. Elution with light petroleum (b.p. 40—60°) gave a solid (0.12 g) which on crystallisation from chloroform-methanol yielded the pure diene (2), m.p. 165—168°, $[\alpha]_{\rm D}$ –20.93° (Found: C, 88.3; H, 11.65. C₂₉H₄₆ requires C, 88.25; H, 11.7%), $v_{\rm max}$. 872 cm⁻¹ (trisubstituted double bond), $\lambda_{\rm max}$. (cyclohexane) 249 (ε 9 100) and 255 nm (11 000), δ 0.75 (3 H, d, J 6 Hz, 4-Me), 0.8—1.28 (18 H, 6 × s, 5-, 13-, 14-, and 17-Me and 20-Me₂), and ca. 5.5 (2 H, m, H-7 and -11), m/e 394 (M⁺), 379, 255, 242, 241, 215, and 205.

25-Hydroxy-D: A-friedo-olean-7-ene (3b).—25-Acetoxy-D: A-friedo-olean-7-ene (3a) (0.23 g) and lithium aluminium hydride (0.4 g) in tetrahydrofuran (40 ml) were refluxed for 6 h. Usual work-up furnished a solid (0.24 g) which was chromatographed over silica gel (20 g). Elution with light petroleum-benzene (7:3, v/v) furnished a solid (0.2 g) which on crystallisation from chloroform-methanol yielded pure alkene (3b), m.p. 192—194°, [α]_D -55.35° (Found: C, 83.65; H, 11.85. C₃₀H₅₀O requires C, 83.45; H, 11.8%), v_{max}. 3 480 (OH) and 860 cm⁻¹ (trisubstituted double bond), δ 0.76 (3 H, d, J 4 Hz, 4-Me), 0.83–1.22 (18 H, 6 \times s, 5-, 13-, 14-, and 17-Me and 20-Me₂), 3.78 (2 H, q, J 12 Hz, H₂-25), and 5.72 (1 H, 2 \times d, J 4 and 7 Hz, H-7), m/e 426 (M^+), 411, 408, 395, 302, 271, 243, 220, 205, and 189.

25-Nor-D: A-friedo-oleana-7,9(11)-diene (2) from (3b).-A mixture of the alcohol (3b) (0.2 g), lead tetra-acetate (1 g) and copper diacetate (0.1 g) in benzene was refluxed in an atmosphere of N₂ for 3 h. The mixture was filtered and worked up as usual to yield a gummy mass (0.2 g) which was chromatographed over silica gel (20 g). Elution with light petroleum furnished a solid (0.11 g) which on crystallisation from chloroform-methanol yielded the pure diene (2), m.p. 165-168°, identical (mixed m.p., t.l.c., rotation, and n.m.r., mass and i.r. spectra) with the diene obtained from putrone.

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